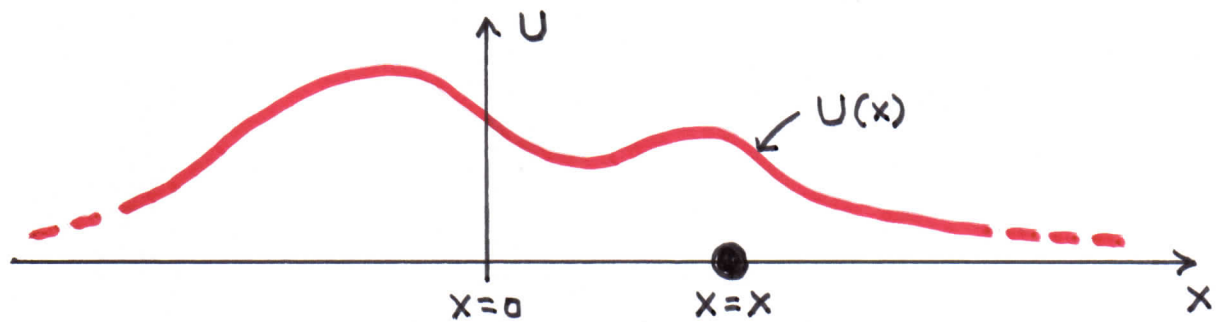


# Preliminary



Consider an isolated system with a single (point mass) particle in a potential energy field  $U(x)$  [for simplicity the system will be 1-D with the particle constrained to move along the  $x$ -axis. In classical mechanics we may be interested in the following quantities:

- (1) particle position vs. time:  $x(t)$
- (2) particle velocity vs. time:  $v(t)$
- (3) kinetic energy vs. time:  $K(t)$
- (4) potential energy vs. time:  $U(x[t])$
- (5) particle momentum vs. time:  $p(t)$
- (6) limits of motion:  $x_{\max}, x_{\min}$  (if any)
- ⋮

# Analysis of a Mechanical System (1-D Classical)

## Method: Integration of Energy Equation

system isolated  
(single mass)

$$\frac{1}{2} m v^2 + U(x) = E = \underline{\text{constant}} \leftarrow$$

$$\frac{1}{2} m \dot{x}^2 + U(x) = E \leftarrow$$

we assume here  
that E is known

$$\dot{x}^2 = \frac{2(E - U(x))}{m}$$

$$\frac{dx}{dt} = \sqrt{\frac{2}{m}} \sqrt{E - U(x)}$$

$$\frac{dx}{\sqrt{E - U(x)}} = \sqrt{\frac{2}{m}} dt$$

$$\int_{x_0}^x \frac{dx}{\sqrt{E - U(x)}} = \sqrt{\frac{2}{m}} \int_0^t dt$$

$$\text{get } t = \sqrt{\frac{m}{2}} \int_{x_0}^x \frac{dx}{\sqrt{E - U(x)}}$$

Invert this and get  $x = f(t)$

$$\text{Then } v(t) = \dot{f}(t)$$

$$K(t) = \frac{1}{2} m [\dot{f}(t)]^2$$

$$E = K(t) + U(x[t])$$

## Alternate method: Solving Differential Equation of N2

$$F = ma \quad \text{note: } F(x) = -\frac{dU(x)}{dx}$$

$$F(x) = m\ddot{x}$$

$$\ddot{x} - \frac{1}{m}F(x) = 0$$

Solve 2nd order O.D.E and get functional solution which includes 2 undetermined constants (call

$$c_1 \text{ and } c_2) \quad x = f(c_1, c_2, t)$$

Find  $c_1$  and  $c_2$  by setting

$$x(t=0) = x_0 \text{ and}$$

$$\dot{x}(t=0) = v_0$$

$$E = K(t) + U(x[t])$$

# Operators

<u>quantity</u>	<u>classical physics</u>	<u>quantum mechanics</u>
position	$x$	$\hat{x} = x$
momentum	$p = mv$	$\hat{p} = -i\hbar d/dx$
K. E.	$K = \frac{1}{2}mv^2$	$\hat{K} = \frac{-\hbar^2}{2m} d^2/dx^2$
P. E.	$U(x)$	$\hat{U} = U(x)$
total E	$E = K + U$	$\hat{E} = \hat{K} + \hat{U}$

If a particle has a well-defined quantity (momentum, total energy, etc.), it will have a wave function for which

operator  $\hat{\Psi}(x) = \text{specific value}$   
of quantity  $\Psi(x)$

eg

$$\hat{p} \Psi(x) = p \Psi(x)$$

$\Psi(x)$  is an eigenfunction of  $\hat{p}$   
 $p$  is an eigenvalue of  $\hat{p}$

# Examples of Operators

eg Consider the differential operator  $\hat{a} = 5 \frac{d}{dx}$

$$\hat{a}(10e^{6x}) = 5 \frac{d}{dx}(10e^{6x}) = 300e^{6x}$$

$$\hat{a}(10e^{6x}) = 30(10e^{6x})$$

this is an  
eigenfunction  
of  $\hat{a}$

↑  
this is an eigenvalue  
of the operator  $\hat{a}$   
associated with the  
eigenfunction  $10e^{6x}$

eg Consider the same operator  
as above  $\hat{a} = 5 \frac{d}{dx}$

$$\hat{a}(x^2 - 3x) = 5 \frac{d}{dx}(x^2 - 3x) = 10x - 15$$

$$\hat{a}(x^2 - 3x) \neq \text{constant}(x^2 - 3x)$$

↑  
this is not an eigen-  
function of operator  $\hat{a}$

# Schrödinger Equation

classical mechanics  $\rightarrow \left. \begin{aligned} \frac{1}{2}mv^2 + U(x) &= E \\ \text{or } \frac{p^2}{2m} + U(x) &= E \end{aligned} \right\} \text{total energy equation}$

quantum mechanics  $\rightarrow \hat{K}\Psi(x) + \hat{U}(x)\Psi(x) = E\Psi(x)$   
or  $\frac{-\hbar^2}{2m} \frac{d^2}{dx^2} \Psi(x) + U(x)\Psi(x) = E\Psi(x)$

this is a second order ordinary D.E.

---

The 3-D version of the Schrödinger Equation:

$$-\frac{\hbar^2}{2m} \left[ \frac{\partial^2 \Psi}{\partial x^2} + \frac{\partial^2 \Psi}{\partial y^2} + \frac{\partial^2 \Psi}{\partial z^2} \right] + U(x, y, z)\Psi = E\Psi$$

where  $\Psi = \Psi(x, y, z)$

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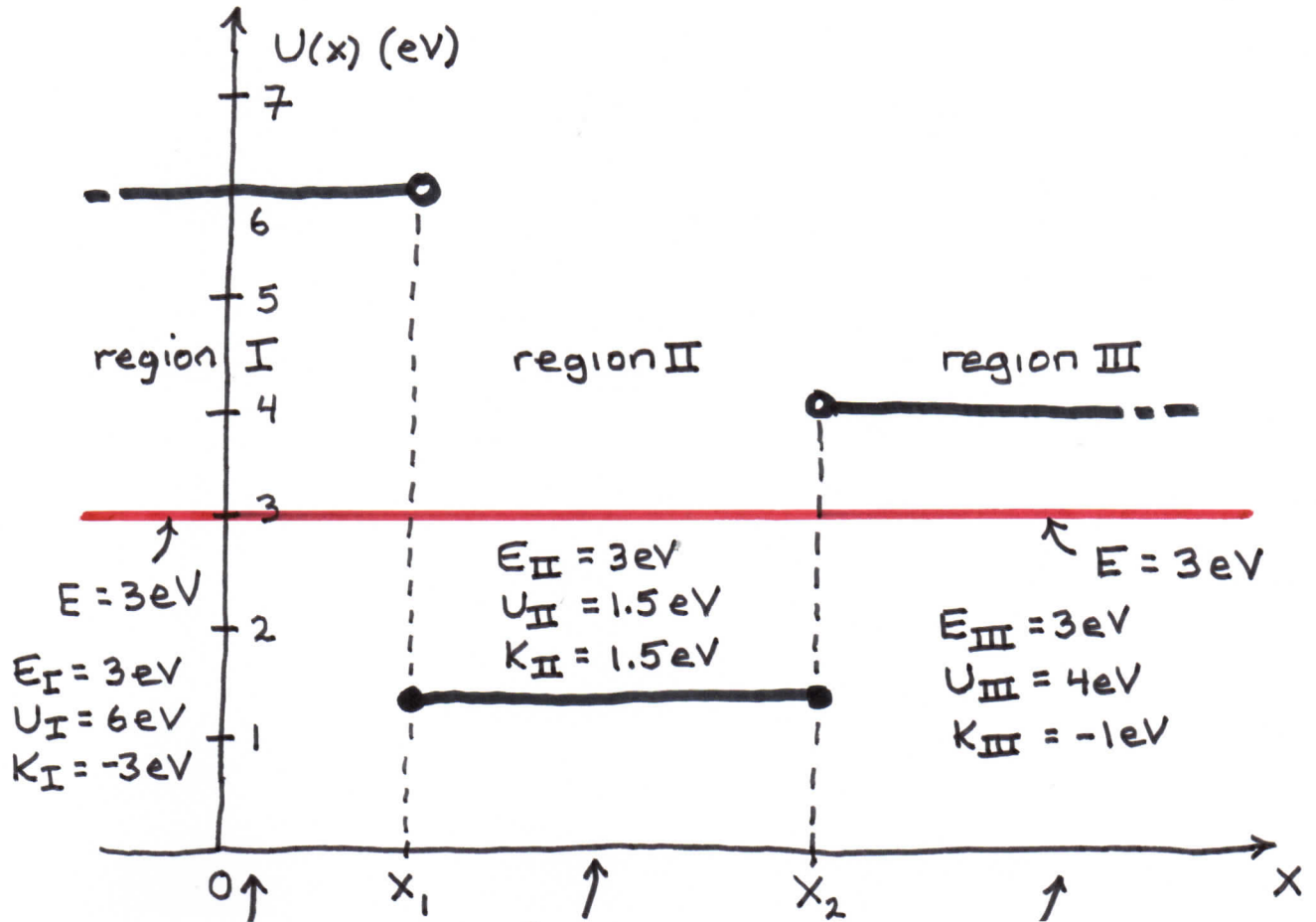
## Erwin Schrödinger

- \* Austrian Physicist
- \* 1926 - paper on wave mechanics, formalizing the Schrödinger Equation and using it to yield correct values for the energy eigenvalues for the hydrogen atom
- \* 1926 - paper solving the quantum harmonic oscillator, rigid rotor, diatomic molecule



Nobel Prize in physics  
1933

# Step Function Potential Energy



$$-\frac{\hbar^2 \psi_I''}{2m} + (6-3)\psi_I = 0 \quad -\frac{\hbar^2 \psi_{II}''}{2m} + (1.5-3)\psi_{II} = 0 \quad -\frac{\hbar^2 \psi_{III}''}{2m} + (4-3)\psi_{III} = 0$$

$$\psi_I = c_1 \exp\left\{-\sqrt{\frac{2m(6-3)}{\hbar^2}} x\right\} + c_2 \exp\left\{+\sqrt{\frac{2m(6-3)}{\hbar^2}} x\right\}$$

$$\psi_{II} = c_3 \sin\left[\sqrt{\frac{2m(3-1.5)}{\hbar^2}} x\right] + c_4 \cos\left[\sqrt{\frac{2m(3-1.5)}{\hbar^2}} x\right]$$

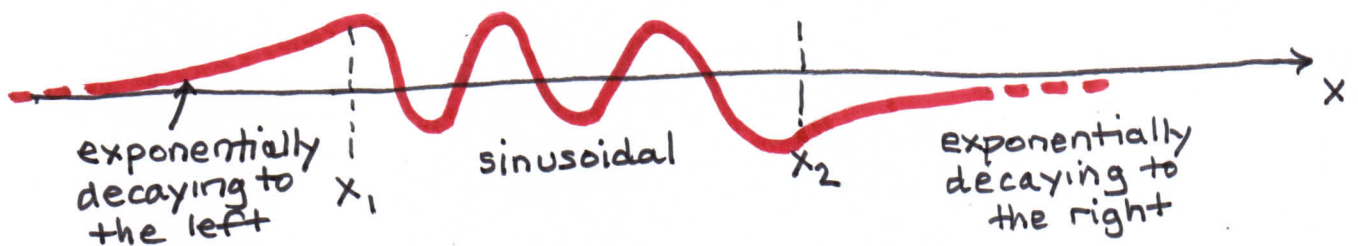
$$\psi_{III} = c_5 \exp\left\{-\sqrt{\frac{2m(4-3)}{\hbar^2}} x\right\} + c_6 \exp\left\{+\sqrt{\frac{2m(4-3)}{\hbar^2}} x\right\}$$

$$\psi_I(x_1) = \psi_{II}(x_1)$$

$$\psi_I'(x_1) = \psi_{II}'(x_1)$$

$$\psi_{II}(x_2) = \psi_{III}(x_2)$$

$$\psi_{II}''(x_2) = \psi_{III}''(x_2)$$



# Card Game

Use a full deck of cards (52 cards, 4 suits, 3 face cards [J, Q, K] in each suit,).

You agree to playing gambling game with the "dealer". He shuffles the cards, and then lets you cut them.

You turn over the card where you've cut the deck and reveal it. Here are the rules:

- ① If the card is a red ace, the dealer pays you \$10.00.
- ② If the card is a black face card, the dealer pays you \$3.00.
- ③ If the card is a numbered spade, the dealer pays you \$2.00.
- ④ If the card is not one of the above, you pay the dealer \$2.00.



# Expected Gain per Play

outcome	gain $g_i$	$P_i$	$P_i g_i$
red ace	+\$10	$\frac{2}{52}$	$+\frac{20}{52}$
black face	+\$3	$\frac{6}{52}$	$+\frac{18}{52}$
numbered spade	+\$2	$\frac{9}{52}$	$+\frac{18}{52}$
other	-\$2	$\frac{35}{52}$	$-\frac{70}{52}$

$$-\frac{14}{52} = -\$0.27$$

<g>

# Expectation Values

The expectation value of a physical quantity  $q$  of a quantum mechanical system whose wave function is  $\Psi(x)$ :

$$\langle q \rangle \equiv \int_{-\infty}^{\infty} \Psi(x)^* \hat{q} \Psi(x) dx$$

$$\langle x \rangle = \int_{-\infty}^{\infty} \Psi(x)^* x \Psi(x) dx$$

← expectation value of the x-position of the particle

$$\langle p \rangle = \int_{-\infty}^{\infty} \Psi(x)^* \left[ -i\hbar \frac{d}{dx} \right] \Psi(x) dx$$

← expectation value of the momentum of the particle

$$\langle K \rangle = \int_{-\infty}^{\infty} \Psi(x)^* \left[ -\frac{\hbar^2}{2m} \frac{d^2}{dx^2} \right] \Psi(x) dx$$

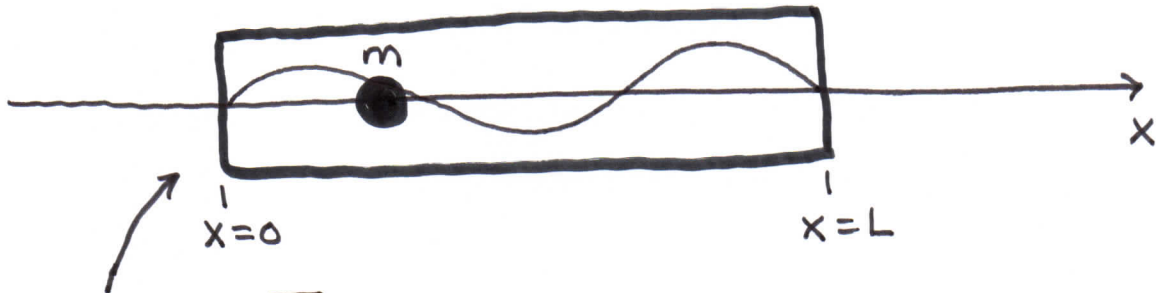
← expectation value of the kinetic energy of the particle

$$\langle E \rangle = \int_{-\infty}^{\infty} \Psi(x)^* \left[ -\frac{\hbar^2}{2m} \frac{d^2}{dx^2} + U(x) \right] \Psi(x) dx$$

← expectation value of the total energy of the system

# Expectation Value Example

## 1-D Particle in a Box:



$$\Psi_3(x) = \sqrt{\frac{2}{L}} \sin\left(\frac{\pi 3x}{L}\right)$$

$$\langle E \rangle = \int_{-\infty}^{\infty} \Psi_3(x)^* \left[ -\frac{\hbar^2}{2m} \frac{d^2}{dx^2} + U(x) \right] \Psi_3(x) dx$$

$$= \int_0^L \sqrt{\frac{2}{L}} \sin\left(\frac{\pi 3x}{L}\right) \left[ -\frac{\hbar^2}{2m} \frac{d^2}{dx^2} + 0 \right] \sqrt{\frac{2}{L}} \sin\left(\frac{\pi 3x}{L}\right) dx$$

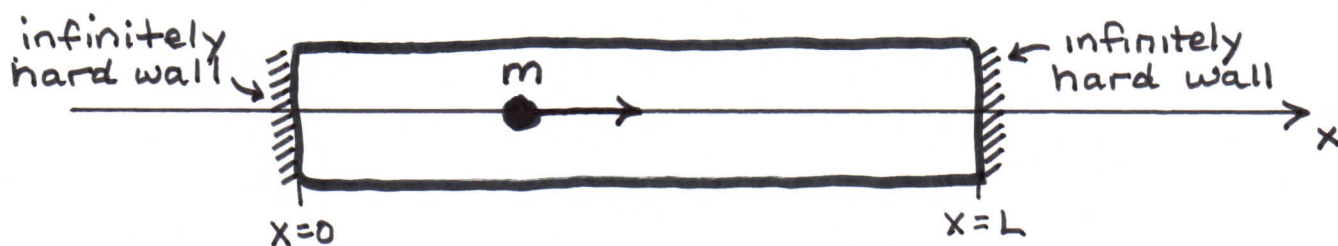
$$= \frac{2}{L} \int_0^L -\frac{\hbar^2}{2m} \left(\frac{\pi 3}{L}\right)^2 (-1) \sin^2\left(\frac{\pi 3x}{L}\right) dx$$

$$= \frac{18 \hbar^2 \pi^2}{2mL^3} \int_0^L \sin^2\left(\frac{\pi 3x}{L}\right) dx = \frac{18 \hbar^2 \pi^2}{2mL^3} \left(\frac{1}{2}L\right)$$

$$= \frac{18 \pi^2}{4mL^2} \left(\frac{h}{2\pi}\right)^2 = \boxed{\frac{3^2 h^2}{8mL^2}}$$

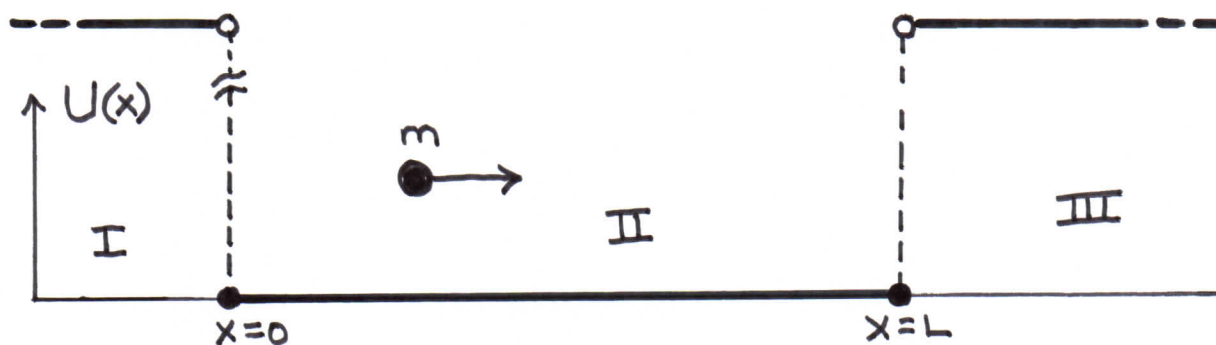
$$\langle x \rangle = \int_{-\infty}^{\infty} \Psi_3(x)^* x \Psi_3(x) dx \longrightarrow \text{can show} = \boxed{\frac{1}{2}L}$$

# Particle in a 1-D Box



\* Wave mechanical approach:  
Find all allowed energies and their associated wave functions.

\* Display potential energy function



region I:  $x < 0$     region II:  $0 \leq x \leq L$     region III:  $x > L$   
 $U_{\text{I}}(x) = \text{infinite}$      $U_{\text{II}}(x) = 0$      $U_{\text{III}}(x) = \text{infinite}$

\* Write Schrödinger Equation for each region

$$\text{region I: } -\frac{\hbar^2}{2m} \psi_{\text{I}}'' + \text{inf. } \psi_{\text{I}} = E \psi_{\text{I}}$$

$$\text{region II: } -\frac{\hbar^2}{2m} \psi_{\text{II}}'' + 0 \psi_{\text{II}} = E \psi_{\text{II}}$$

$$\text{region III: } -\frac{\hbar^2}{2m} \psi_{\text{III}}'' + \text{inf. } \psi_{\text{III}} = E \psi_{\text{III}}$$

$\Psi_I(x)$  must be 0 and  $\Psi_{III}(x)$  must be 0

\* Solve  $-\frac{\hbar^2}{2m} \Psi_{II}'' = E \Psi_{II}$  undetermined  
(as of yet) constant

$$-\frac{\hbar^2}{2m} \Psi_{II}'' - E \Psi_{II} = 0$$

$$\Psi_{II}'' + \frac{2mE}{\hbar^2} \Psi_{II} = 0$$

General solution to the above is:

$$\Psi_{II}(x) = c_1 \sin \sqrt{\frac{2mE}{\hbar^2}} x + c_2 \cos \sqrt{\frac{2mE}{\hbar^2}} x$$

Apply the boundary condition  $\Psi_I(0) = 0 = \Psi_{II}(0)$

$$0 = c_1 \sin \sqrt{\frac{2mE}{\hbar^2}}(0) + c_2 \cos \sqrt{\frac{2mE}{\hbar^2}}(0)$$

$$\boxed{0 = c_2}$$

So  $\Psi_{II}(x) = c_1 \sin \sqrt{\frac{2mE}{\hbar^2}} x$

Apply the boundary condition  $\Psi_{II}(L) = \Psi_{III}(L) = 0$

$$0 = c_1 \sin \sqrt{\frac{2mE}{\hbar^2}} L \quad \leftarrow c_1 \text{ cannot be 0 or you would have no wave function at all}$$

↑  
this will be 0

$$\text{if } \sqrt{\frac{2mE}{\hbar^2}} L = n\pi \quad \leftarrow \text{any integer } 1, 2, 3, 4, \dots$$

$$\sqrt{\frac{2mE}{\hbar^2}} L = n\pi \rightarrow E = \frac{n^2 \pi^2 \hbar^2}{2mL^2} \text{ or } \boxed{\frac{n^2 \hbar^2}{8mL^2} = E_n}$$

These are the allowed energies →

So  $\Psi_{II}(x) = c_1 \sin \sqrt{\frac{2mE}{\hbar^2}} x$  ← synthesize with  $E = \frac{n^2 \hbar^2}{8mL^2}$

→  $\Psi_{nII}(x) = c_1 \sin\left(\frac{n\pi x}{L}\right)$

This must be normalized and so demand

$$\int_{-\infty}^{\infty} |\Psi(x)|^2 dx = 1$$

$$\int_{-\infty}^0 |\Psi_I(x)|^2 dx + \int_0^L c_1^2 \sin^2\left(\frac{n\pi x}{L}\right) dx + \int_L^{\infty} |\Psi_{III}(x)|^2 dx = 1$$

$$c_1^2 \int_0^L \sin^2\left(\frac{n\pi x}{L}\right) dx = 1$$

$$c_1 = \sqrt{\frac{2}{L}}$$

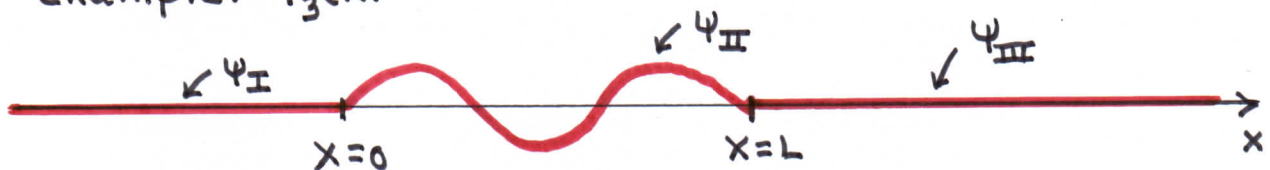
So  $\Psi_{II}(x) \rightarrow \Psi_n(x) = \sqrt{\frac{2}{L}} \sin\left(\frac{n\pi x}{L}\right)$

allowed wave functions of the particle in a 1-D box

$$E_n = \frac{n^2 \hbar^2}{8mL^2}$$

allowed energies associated with those quantum status

example:  $\Psi_3(x)$



$$\hat{E} = \hat{K} + \hat{U} = -\frac{\hbar^2}{2m} \frac{d^2}{dx^2} + 0$$

↑ total energy operator    ↑ kinetic energy operator    ↑ potential energy operator    ↑ inside box

$$\hat{E}\psi = E\psi \leftarrow \text{solve}$$

results for particle in a 1-D box

$$\psi_1 = \sqrt{2/L} \sin\left(\frac{\pi}{L}x\right) \longleftrightarrow E_1 = \frac{h^2}{8mL^2}$$

$$\psi_2 = \sqrt{2/L} \sin\left(\frac{2\pi}{L}x\right) \longleftrightarrow E_2 = \frac{2^2 h^2}{8mL^2}$$

$$\psi_3 = \sqrt{2/L} \sin\left(\frac{3\pi}{L}x\right) \longleftrightarrow E_3 = \frac{3^2 h^2}{8mL^2}$$

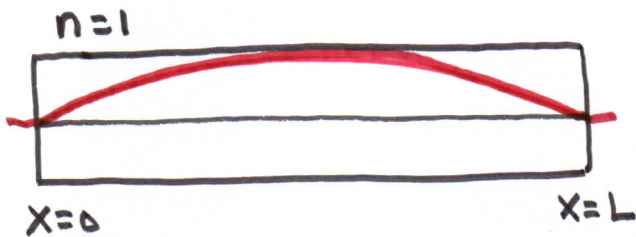
↑ ⋮

these are eigenfunctions of the total energy operator  $-\frac{\hbar^2}{2m} \frac{d^2}{dx^2}$

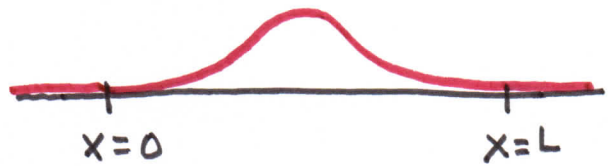
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these are the eigenvalues of the total energy operator  $-\frac{\hbar^2}{2m} \frac{d^2}{dx^2}$  associated w/ the eigenfunctions  $\psi_n$

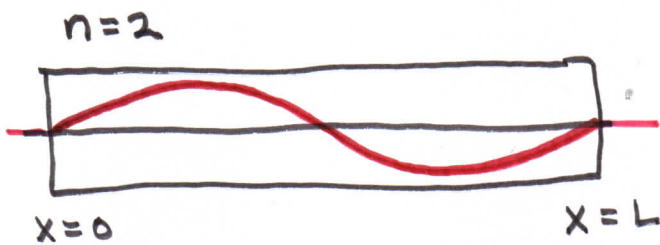
# Particle in a 1-D Box



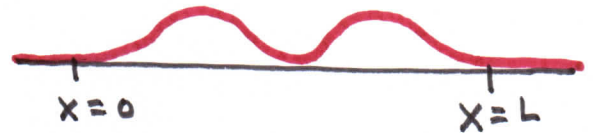
$$\psi_1(x) = \sqrt{\frac{2}{L}} \sin\left(\frac{\pi x}{L}\right)$$



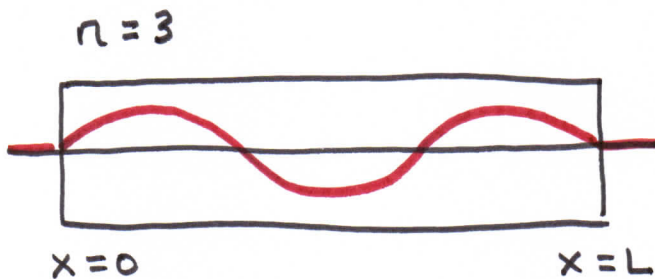
$$|\psi_1(x)|^2 = \frac{2}{L} \sin^2\left(\frac{\pi x}{L}\right)$$



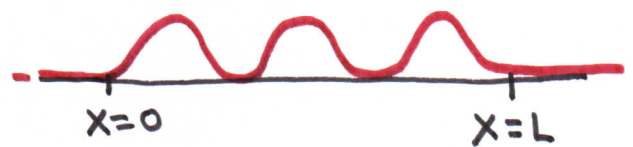
$$\psi_2(x) = \sqrt{\frac{2}{L}} \sin\left(\frac{2\pi x}{L}\right)$$



$$|\psi_2(x)|^2 = \frac{2}{L} \sin^2\left(\frac{2\pi x}{L}\right)$$



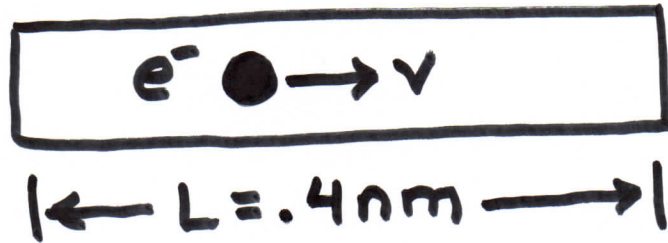
$$\psi_3(x) = \sqrt{\frac{2}{L}} \sin\left(\frac{3\pi x}{L}\right)$$



$$|\psi_3(x)|^2 = \frac{2}{L} \sin^2\left(\frac{3\pi x}{L}\right)$$



# Particle in a 1-D Box



$$E_n = \frac{n^2 h^2}{8 m_e L^2} \quad n = 1, 2, 3, \dots$$

$$\text{Use } E_n = \frac{n^2 (hc)^2}{8 m_e c^2 L^2}$$

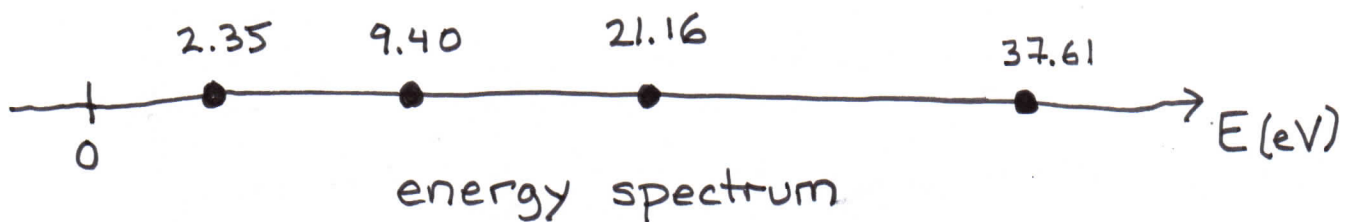
$$E_1 = \frac{1^2 (1240)^2}{8 (511,000) (.4)^2} = 2.35 \text{ eV}$$

$$E_2 = \frac{2^2 (1240)^2}{8 (511,000) (.4)^2} = 9.40 \text{ eV}$$

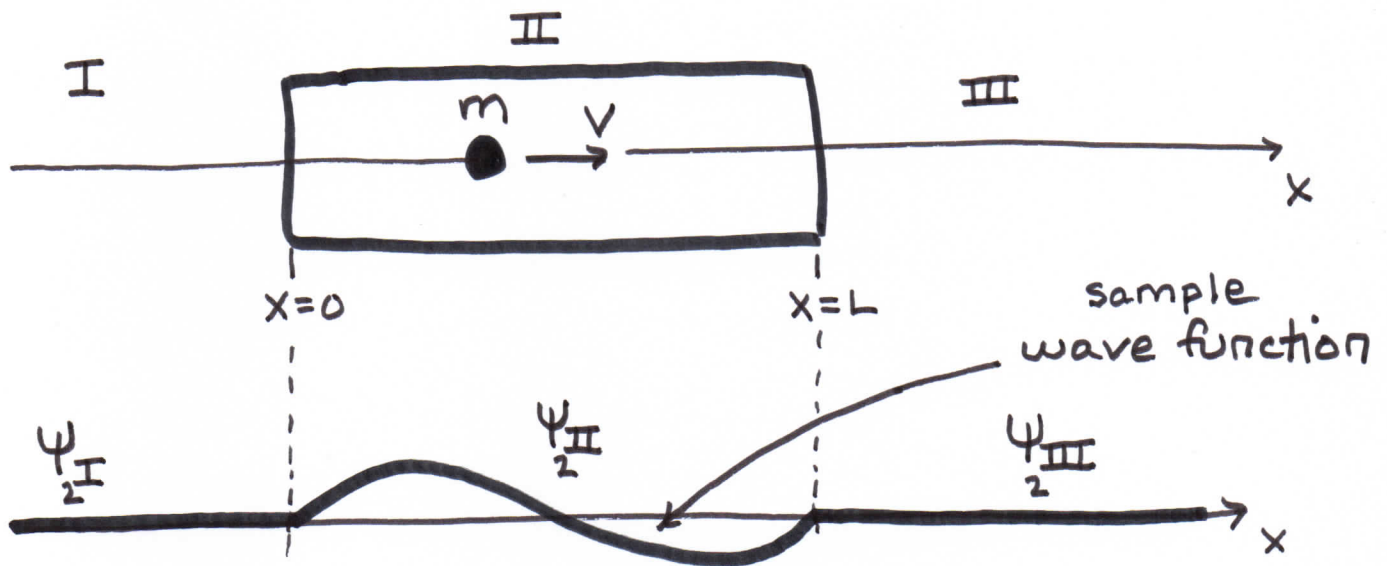
$$E_3 = 21.16 \text{ eV}$$

$$E_4 = 37.61 \text{ eV}$$

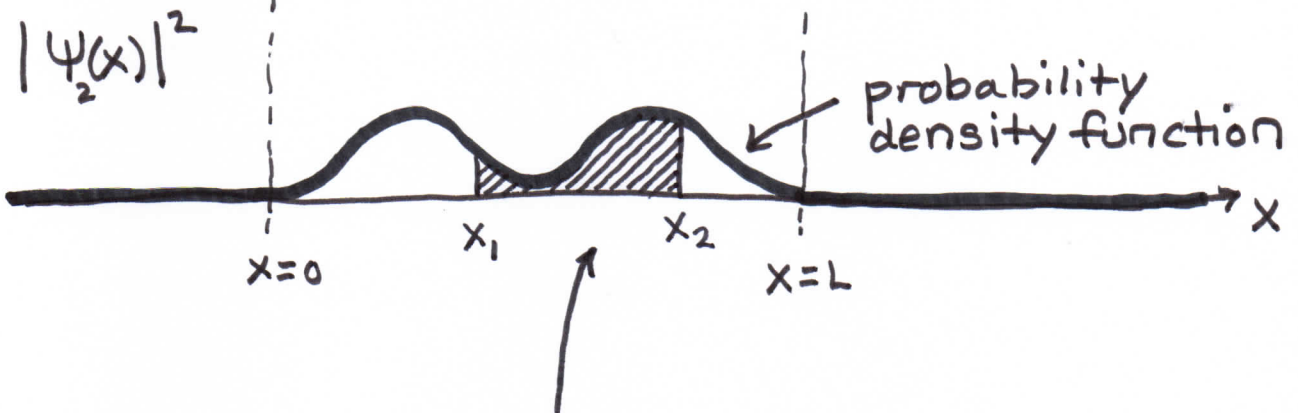
⋮



# Probability Density



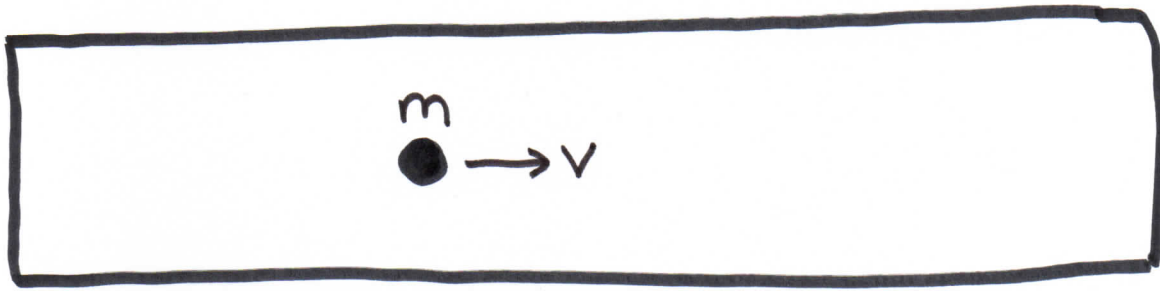
$$\psi_2(x) = \begin{cases} 0 & \text{for } x \leq 0 \\ \sqrt{2/L} \sin\left(\frac{\pi 2x}{L}\right) & \text{for } 0 \leq x \leq L \\ 0 & \text{for } x \geq L \end{cases}$$



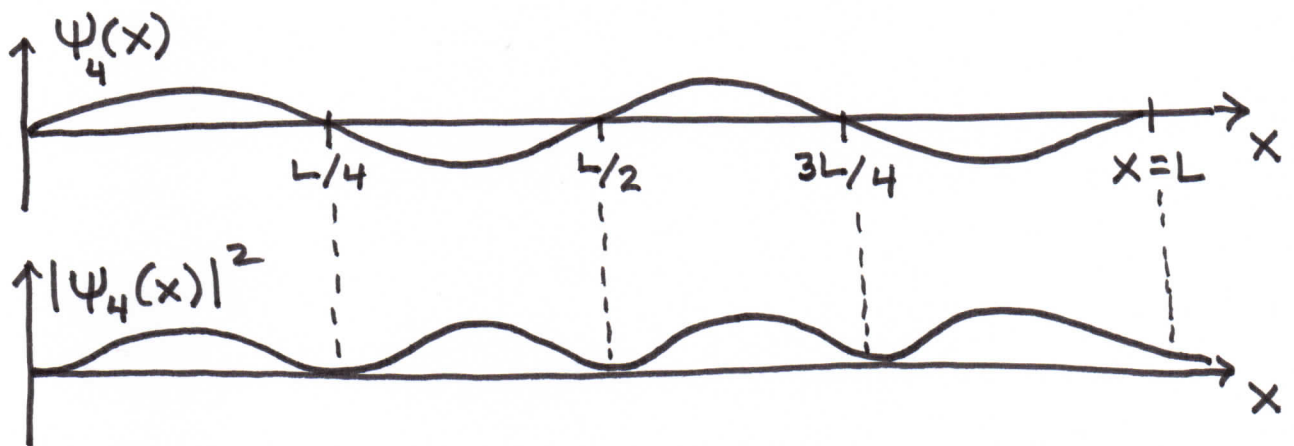
Probability of detecting the particle (with the above wave function) between  $x = x_1$  and  $x = x_2$  is

$$P(x_1 \leq x \leq x_2) = \int_{x_1}^{x_2} |\psi_2(x)|^2 dx$$

# 1-D Box Example



$n = 4$  (3<sup>rd</sup> excited state)



What is probability that particle would be found between  $x = \frac{L}{4}$  and  $\frac{L}{2}$

$$\Psi_4(x) = \sqrt{\frac{2}{L}} \sin\left(\frac{4\pi x}{L}\right)$$

$$P\left(\frac{L}{4} < x < \frac{L}{2}\right) = \int_{L/4}^{L/2} \left(\frac{2}{L}\right) \sin^2\left(\frac{4\pi x}{L}\right) dx$$

$$= \boxed{.25 \text{ or } 25\%}$$